THE

BOTANICAL GAZETTE

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CHEMICAL CONSTITUENTS OF AMARANTHUS RETROFLEXUS

CONTRIBUTIONS FROM THE HULL BOTANICAL LABORATORY 254

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(WITH ELEVEN FIGURES)

Introduction

It is a well known fact that weeds retard the development of cultural plants. This is due to a number of causes: use of water. shading, use of nutrient salts, etc. It has been claimed for various species of Amaranthus that they not only absorb nitrates to care for their nutrient needs, but that they store much nitrogen as nitrate. If this be true, this genus has an excellent adaptation to enable it to combat cultural plants, for nitrate supply is a common limiting factor for crop growth. In order to investigate this statement, to locate the place of nitrate storage, and to determine the amount of nitrogen used otherwise by this plant, separate analyses were made of roots, stems, leaves, and branches of Amaranthus retroflexus at various stages and under various conditions of growth. The amount of the several carbohydrates was also determined in each analysis, in order to calculate the carbohydrate-nitrogen ratio which is lately receiving so much attention. A tissue analysis of the seeds was also made in an endeavor to ascertain more fully the chemical constituency of this plant, with the hope of learning more of the peculiar germinative behavior of these seeds.

Historical

The first chemical analysis of Amaranthus was made by BOUTIN (1), a French chemist, in 1873. Housewives of that time were using these plants to clean their cooking utensils. This fact gave it some commercial importance and brought it to the attention of chemists. It was thought that the ability to "cut grease" must be due to acids in the plant. To verify this BOUTIN incinerated 100 gm. dry weight of the entire plant, and obtained 16 gm. residue. Water was added to leach out the soluble salts. The soluble portion weighed 8 gm. He called this potassium carbonate, and calculated the equivalent weight in grams of potassium nitrate, and found it to be 11.68, or 11.68 per cent. Boutin concluded that this plant was neutral in reaction on account of the presence of the neutral salt KNO₃ (as a matter of fact this plant is acid in reaction). Later (2) he made analyses of the other species of Amaranthus by the same method, to determine the amount of KNO₃. The result of his analysis was as follows: A. atropurpureus contains 22.77 per cent KNO₃ (one kilogram gives 31 gm. N and 103.5 gm. K); A. Blitum contains 11.68 per cent KNO₃; A. ruber contains 16 per cent KNO₃ (one kilogram gives 22 gm. N and 72 gm. K). It is evident that Boutin's method is not an accurate quantitative method of determining the amount of nitrate present. On the other hand, he demonstrated by other qualitative methods that the several species of Amaranthus studied contain a large amount of potassium nitrate. Brosset (3) suggests the use of these plants as a fertilizer.

PAMMEL and Dox (15), in 1917, made microchemical tests of three common pigweeds, A. blitoides, A. graecizans, and A. retro-flexus, and found them to contain abundant starch, some protein, and a little fat. In addition they made the Kjeldahl-Gunning nitrogen determination and found these species to have 1.88, 2.32, and 2.49 per cent of nitrogen. Multiplying by the factor 6.25, they obtained 11.75, 14.52, and 15.59 per cent of protein respectively.

Harding and Egge (8) made an analysis of the seeds of A. retroflexus for fats, protein, starch, sugars, hemicellulose, crude fiber, and tannin.

Of late much significance is being attached to the carbohydratenitrogen ratio of plant tissues, or, as Fischer (6) puts it, the $\frac{C}{N}$. On the basis of work done by him and others, FISCHER makes the following generalizations. If the value of $\frac{C}{N}$ rises by an increase in the amount of carbon, or by a decrease in the amount of nitrogen furnished the plant, there is an increase in the amount of flowering. If the value of $\frac{C}{N}$ drops by a decrease in the amount of carbon, or by an increase in the amount of nitrogen furnished the plant, there is an increase of vegetative growth and a reduction of flowering. Briefly stated, a great preponderance of carbohydrates in plants favors flowering. Since the carbon of plants is fixed from the carbon dioxide of the air by photosynthesis, conditions that favor photosynthesis will tend to increase the ratio, and according to FISCHER the flower production. He found that increased partial pressures of carbon dioxide in the air had this effect. Since nitrogen is absorbed from the soil in the form of nitrates, conditions that favor nitrate absorption will decrease the ratio, and according to Fischer favor vegetation.

Kraus and Kraybill (12), on the basis of much more critical work, including numerous cultures, tissue analyses, and microchemical and anatomical studies, conclude that a very high carbohydrate value gives little vegetation and little or no reproduction; a medium $\frac{\text{carbohydrate}}{N} \quad \text{value gives moderate vegetation}$ and good reproduction; and a low $\frac{\text{carbohydrate}}{N} \quad \text{value gives vigorous vegetation and little reproduction.}$ Through their extreme conditions of culture, withholding nitrates, it is probable that Kraus and Kraybill got much higher carbohydrate plants than Fischer obtained in his cultures, hence their conclusion that very high $\frac{\text{carbohydrate}}{N} \quad \text{gives little vegetation or reproduction.} \quad \text{In short,}$ Fischer worked only on the portion of the $\frac{C}{N} \quad \text{curve that induced}$

fair vegetation and good reproduction or extreme vegetation and little reproduction, but not on the extreme of the curve that greatly reduces both vegetation and reproduction. Kraus and Kraybill cite literature showing that various conditions that greatly retard growth produce high carbohydrate plants. It seems that such conditions retard the use of carbohydrates for building new tissue to a greater degree than they do photosynthesis, and thereby lead to an accumulation of carbohydrates.

Hedlund (9) finds that under like cultural conditions those varieties of winter wheat that have a higher percentage dry weight in the autumn are generally more winter hardy than the ones having a low percentage dry weight, and that cultural conditions that make for high percentage dry weight in any variety also make for winter hardiness. He finds, as do Kraus and Kraybill, that high percentage dry weight is due to high percentage carbohydrate, and therefore high $\frac{\text{carbohydrate}}{N}$.

RIBERA (16) finds that all cultural conditions that increase the percentage dry weight in wheat decrease lodging. From this and the two investigations previously mentioned it is evident that carbohydrate

high $\frac{\text{carbohydrate}}{N}$ increases straw strength and decreases lodging.

High percentage of carbohydrate is said to increase hardiness, at least in part, by the greater amount of glucose present, and it may increase straw strength by inducing greater development of mechanical tissue along with greater thickness of walls, as Kraus and Kraybill found for certain tissues of the tomato.

Methods and results

GREEN PLANT AT VARIOUS STAGES OF DEVELOPMENT

Preparation of samples.—Samples were secured on June 3, June 20, and July 8 consecutively from a vacant lot on 59th Street and Ingleside Avenue, Chicago. On June 20 samples were taken from two places, namely, the manure pile (rich soil) and the knoll (poor soil) for comparative work. The soil particles adhering to the roots and rootlets were removed by running water from a filter pump. As the velocity of water was very great, the soil particles were removed without difficulty. The roots were partially dried by the air current from the laboratory air line, and

finally dried by the use of paper towels. The roots, stems, and leaves were detached for separate analysis. The roots and the main stems were separated by cutting just between the cotyledon scar and the first branching rootlet. The leaf blades with the petioles were separated from the stem. Each portion was weighed and the length and diameter measured. Table I gives a brief description of the three consecutive samples.

TABLE I

THE GREEN PLANT AT VARIOUS STAGES OF GROWTH

Measurement	June 3 collection	June 20 collection	July 8 collection
9	Inches	Inches	Inches
Average height	1.00-4.00	6.00-8.00	20.00
Taproot { length	1.00-1.50	4.00-6.00	4.00-6.00
diameter		0.10-1.00	1.00-1.20
Secondary rootlets { length diameter	None	0.10-1.00	8.00-14.00
diameter	None		0.13-0.25
Stome Slength	1.00-4.00	6.00-8.00	20.00
Stems {lengthdiameter	0.13-0.25	0.37-0.75	0.13-0.25
Lateral branches { length diameter	None	0.10-1.50	14.00
diameter	None		0.25-0.50
Seed head	None	None	0.25-1.00
Green weight	Grams	Grams	Grams
Roots	26.00	28.50	29.20
Main stems	52.05	97.40	88.30
Branches	None	None	80.45
Leaves	188.90	142.20	110.25

The green samples were then immediately put in a freezing chamber, allowed to freeze overnight, and ground in a meat grinder the next morning. The freezing prevents losses. In the frozen condition no juices ooze out or spatter in the manipulation. The samples were boiled with 95 per cent alcohol to destroy the enzymes, and were then transferred to extraction cups, with filter paper thimbles, previously dried and weighed. The-tissues were fractionated according to Koch's (II) scheme for tissue analysis, namely, the lipin or ether soluble fraction (F_I), the alcohol water soluble fraction (F_I), and the insoluble fraction (F_I). In the green plant F_I was comparatively small, consisting of chlorophyll and extracts of various pigments. The F_I was put together with F_I for the following carbohydrate and nitrogen estimation.

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ANALYSES

Nitrogen compounds

NITRATES.—The nitrates were determined by the Schlösing-Wagner method (14) as modified by Koch for use in his laboratory. The modification consists essentially in the use of an inverted burette instead of a tube sealed at one end, and of the Van Slyke apparatus (only volumetric tube and Hemple pipette), to measure the true volume of nitric oxide. The principle of the method is: ${}_{3}\text{Fe}^{++} + \text{NO}_{3}^{-} + {}_{4}\text{H}^{+} \rightarrow {}_{3}\text{Fe}^{+++} + \text{NO}\,\text{gas} + {}_{2}\text{H}_{2}\text{O}$. Therefore I mol. NO gas gives an equivalent of 62 gm. of NO₃, or I cc. of ${}_{3}\text{gas} = \frac{62}{22400} = 2.77 \text{ mg. NO}_{3}$.

In order to determine the accuracy of this method, a known solution of KNO₃ (0.5 per cent) was used. Four consecutive determinations with 10 cc. of the known solution were made. The average volume of nitric oxide gas for each 10 cc. solution, calculated to standard condition, was 11.12 cc. The theoretical volume for 10 cc. of 0.5 per cent KNO₃ is 11.078 cc.

The determination of nitrates in the samples was made by taking an aliquot of the soluble fractions (F_1 and F_2). The nitric oxide gas driven over was caught in an inverted burette which had previously been filled with 40 per cent NaOH to absorb the CO_2 and neutralize the hydrochloric acid (HCl gas will come over when the hydrochloric acid concentration reaches 20 per cent in the boiling flask). The burette containing the nitric oxide gas was set aside and allowed to cool to room temperature; then the nitric oxide gas was transferred to the Van Slyke apparatus. The total volume and the volume of unabsorbed gas were recorded. The absorbed volume by the alkaline KMnO₄ in Hemple pipette is that of nitric oxide. This volume was then calculated to standard volume from temperature and barometric pressure, for example:

	1	11
Aliquot in cc. used	25.0	25.0
Total volume of gas (nitric oxide+air)	4.80	4.85
Volume of unabsorbed gas (air)	1.27	I.26
Volume of (absorbed) nitric oxide	3 · 53	3 · 59
Barometric pressure = 746.9; temperature 20.5° C.		
Volume at standard condition	3.23	3.27

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	1	П
Equivalent in milligrams of NO ₃	8.95	9.06
Milligrams of dry substance (25 cc.) used	107.75	107.75
Percentage NO ₃ in soluble fractions F ₁ +F ₂	8.30	8.41
The percentage soluble fractions in whole samples	45.9	45.9
Therefore percentage of NO ₃ calculated on whole sample	3.81	3.86

Soil samples were taken at the same time that the green plants were gathered. The nitrates were estimated by the colorimetric method with phenoldisulphonic acid. The moisture in percentage and nitrates in parts per million are shown in fig. 1.

On June 20 the samples from the manure pile and from the knoll were taken for comparison. The nitrate content of the soil and that in the plant were as follows:

	Knoll sample	Manure pile sample
NO ₃ in ppm in soil	300	29
NO ₃ in percentage in plant (stem only)	1.71	1.45

The high NO₃ content in the soil of the knoll sample is probably due to the fact that some one, perhaps the gardener, had disturbed the soil by dragging his cultivator over it accidentally in cultivating his plot near by. The second reason is the better drainage and aëration in the knoll, and therefore better conditions for nitrification; but the striking fact is that high nitrate content in the soil did not bring about a proportional high nitrate content in the green plant organs. The rate of absorption increases with the aging of the plant; when the plants were about 25 days old, the nitrate in the stem was only 1.71 per cent. Eighteen days later (July 8) the nitrate content had risen to 8.58 per cent. During the same period branches grew from 0.1 to 14 inches, and their nitrate content rose to 12.50 per cent. This rapid increase in absorption of nitrates may partially be explained by the increase in extent of the absorbing roots from a radius of a few inches to about 2 ft.

The nitrate content in the roots, stems, and leaves is given in table II and is also shown in fig. 1. The nitrate content of the roots falls gradually from 1.85 per cent on June 3 to zero on June 20. At the same time nitrates in the leaves fall from 1.38 per cent to zero, while in the stem there is a gradual increase. There must be a definite reason for such differences. The differences may be due

to many causes. In the leaves protein synthesis is going on continuously in the presence of soluble carbohydrates. There is also synthesis of other organic nitrogen compounds, such as chlorophyll,

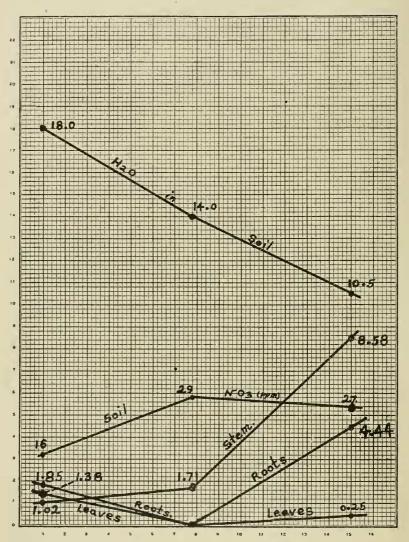


Fig. 1.—Relation of soil nitrification and nitrate intake by green plant organs; nitrate in soil expressed in parts per million; all other data calculated as percentage on dry weight basis.

phospholipins, etc., and all of the nitrates of the leaves seem to be used up in these syntheses. The nitrates are carried from the roots to other parts of the plants as fast as they are taken up from the soil. There may be as high a concentration of nitrates in the roots as in the soil (29 parts per million). This low concentration cannot be estimated by this method and would therefore be missed.

The stem and branches are the primary nitrate storage organs. The nitrate content rises as high as 8.58 per cent in the stem and 12.5 per cent in the branches during the early seed formation period. This high content is shown still more clearly by the ratio of nitrate nitrogen to the total nitrogen. This is 32.8 per cent for the roots, 51.85 per cent for the stems, 56.4 per cent for the branches, but only 1.25 per cent in the leaves. Curves showing this ratio in these organs at different stages are given in fig. 2. This large supply of nitrates in the stem and branches may be drawn upon heavily for further growth and seed production, although the supply seems more than adequate for these uses. There is also no reason for thinking that nitrate absorption ceases at this time. The extent to which this storage of nitrate is drawn upon by later development could be ascertained by the analysis of a set of samples taken late in the fall when seed formation and growth were complete. It is to be regretted that circumstances made such an analysis impossible for this paper.

It is worthy of note that the nitrate storage organs are the ones that made the most rapid growth in length, weight, and volume. The stem which rose from 8 inches on June 20 to 20 inches on July 8 at the same time increased in nitrate content from 1.71 to 8.58 per cent on dry weight basis. In addition to the stem there are numerous side branches which elongated from 0.1 to 14 inches in 18 days, making nearly 1 inch in 24 hours. At the same time there was an increase in percentage of nitrates per gram of dry matter from 1.71 on June 20 to 12.50 per cent on July 8. The rate of nitrate intake per gram per hour seems to follow a geometrical progression in each individual plant.

It appears that *Amaranthus* may be a very considerable factor in depleting soils of their nitrates. Also in case the weeds are burned the nitrogen stored is permanently lost from the soil. It

is a point of interest to know how generally this great power of absorbing and storing nitrates is possessed by weeds.

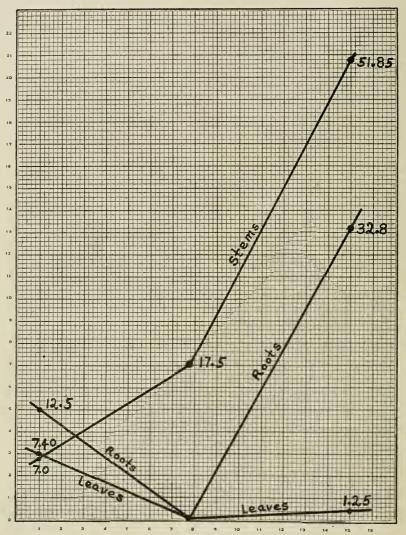


Fig. 2.—Ratio of nitrate nitrogen to total nitrogen; ratio for branches (not shown in figure) 56.4.

Amino N.—The amino nitrogen was determined by the Van Slyke apparatus. The amino acids thus determined are chiefly

of the mono-amino-monocarboxylic acids. In each estimation only 2 cc. of the solution was used. The amino acid nitrogen deter-

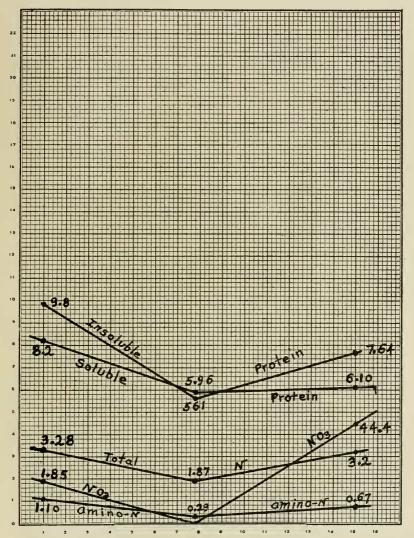


Fig. 3.—Nitrogen compounds in roots

mined throughout the season is given in table II. Curves showing the variation in roots, stems, and leaves are given in figs. 3, 4, and 5 respectively. In all the three organs the variation is very small. In general the amino N varies directly with the insoluble protein; with a high protein content, high amino N; and low protein content, low amino N.

TABLE II

THE NITROGEN COMPOUNDS IN THE GREEN PLANT

	P .	C	Branches	Leaves					
24	Roots	Stems	Branches	Leaves					
Material		June 3 collection	on (1-4 inches)						
Total N	3.33 3.33	3.19 3.21		4.25 4.27					
Nitrates NO ₃	1.77 1.93	0.99 1.04		1.36 1.40					
Nitrate N	0.40 0.43	0.22 0.23		0.31 0.32					
Amino N	1.09 1.13	0.43 0.43		0.97 0.96					
Insoluble N	1.62 1.52	1.74 1.73		3.21 3.23					
Insoluble protein	10.18 10.25	10.92 10.80		20.20 20.30					
Soluble protein	8.24 8.17	7.67 7.92	• • • • • • • • • •	4.58 4.52					
		June 20 collection (6–8 inches)							
Total N	1.85 1.88	2.39 2.21		4.56 4.51					
Nitrates NO ₃	None None	1.67 1.75		None None					
Nitrate N	None None	0.38 0.40		None None					
Amino N	0.29 0.29	0.19 0.19		1.38 1.37					
Insoluble N	0.86 0.91	0.95 0.97		3.66 3.59					
Insoluble protein	5.40 5.72	5.86 6.10		23.00 22.6					
Soluble protein	5.92 6.10	5.28 5.41		5.66 5.78					
		July 8 collecti	ion (20 inches)						
Total N	3.21 3.12	3.74 3.80	5.04 4.94	4.85 4.80					
Nitrates NO ₃	4.41 4.47	8.40 8.75	12.50 12.40	0.25 0.246					
Nitrate N	0.99 1.02	1.90 1.98	2.85 2.80	0.06 0.06					
Amino N	0.63 0.70	0.35 0.34	0.48 0.49	1.44 1.42					
Insoluble N	1.29 1.14	0.95 0.91	1.01 0.98	3.29 3.21					
Insoluble protein	8.10 7.17	5.96 5.71	6.35 6.15	20.62 20.08					
Soluble protein	6.16 6.05	5.60 5.70	7.42 7.30	9.42 9.62					

Insoluble protein.—The insoluble protein was calculated from nitrogen of the insoluble fraction (F₃). The insoluble protein is given in table II, and curves showing fluctuation during the growing season are given in figs. 3–5. The insoluble protein falls and then rises again at maturity in the root, while in the leaves the fluctuation is in the opposite direction (see curves). In the stem the decline is in the early stage from 10.89 per cent (June 3) to 6.03 per cent (June 20). From that time on the curve is almost a straight horizontal line; therefore the rate of synthesis of the insoluble protein must have been keeping pace with the growth of the stem,

because the fall at this time is only from 6.03 to 5.84 per cent (almost within experimental error).

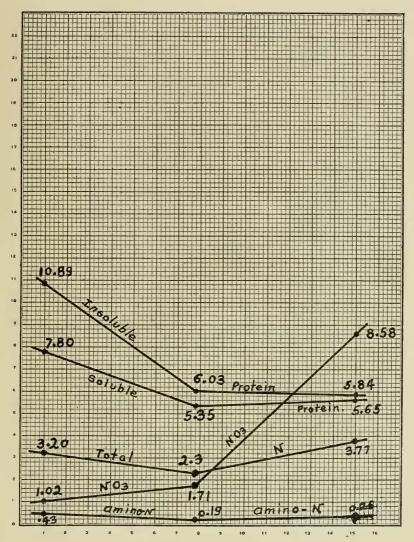


Fig. 4.-Nitrogen compounds in stems

Soluble protein.—Soluble protein in F₁ and F₂ is computed from the Kjeldahl nitrogen determination. The nitrate nitrogen

was determined separately (by the method previously described); so the Kjeldahl determination was conducted without any modifica-

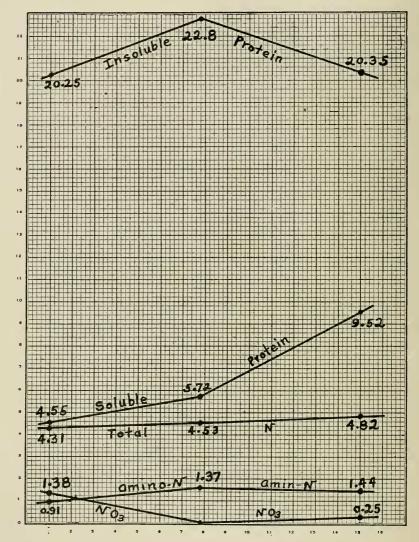


Fig. 5.—Nitrogen compounds in leaves

tion for nitrates. (Previously zinc was used, but this did not reduce any nitrate. Experiment with a known solution 0.5 per cent KNO₃,

per official method by the use of salicylic acid and sodium thiosulphate, reduced only 60 per cent of the nitrate into ammonia.)

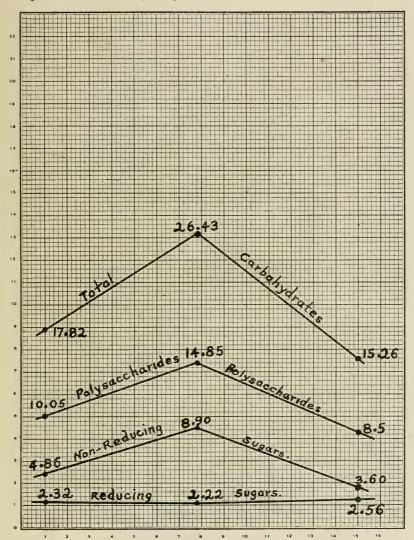


Fig. 6.—Different carbohydrates in roots

It is incorrect to call all this nitrogen as derived from protein, because part of the soluble nitrogen was from the breaking down of the chlorophyll; but for comparison it is not out of place to calculate the N by the factor 6.29 to convert it into soluble protein

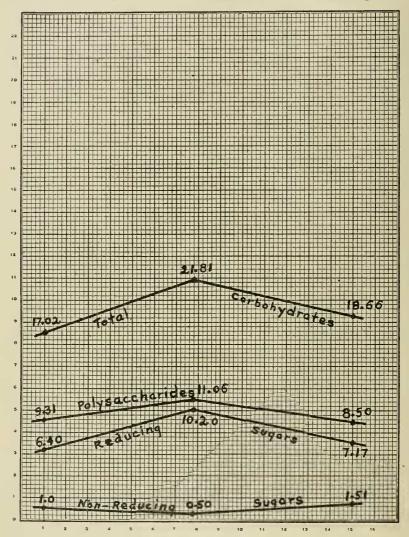


Fig. 7.—Different carbohydrates in stems

for temporary convenience in interpreting the results. The curves of the soluble protein in figs. 3–5 are self-explanatory, showing the variation throughout the season.

CARBOHYDRATES.—The carbohydrates were determined by the reduction method with Fehling solutions. The cuprous oxide

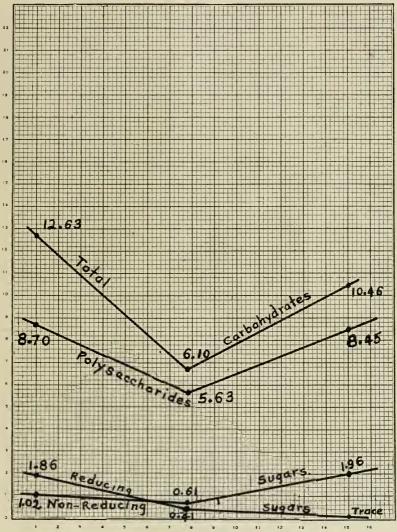


Fig. 8.—Different carbohydrates in leaves

obtained was dissolved in excess of ferric ammonium sulphate with $\rm H_2SO_4$ previously added. The ferrous ions produced by the oxidation of cuprous oxide were titrated against a $\rm N/2o~KMnO_4$

solution, in which i cc. represents 3.1 mg. of copper. The corresponding equivalents of the different sugars expressed in milligrams were found in the Munson-Walker table. The weight of sugar found divided by the material used gives the amount of sugar contained in i gm. of material.

The soluble carbohydrates are in F_1 and F_2 . The reducing sugar was first determined. The non-reducing sugar was obtained by subtracting the reducing sugar from the total sugar by hydrochloric acid hydrolysis at $67-69^{\circ}$ C. for 10 minutes.

The insoluble carbohydrates are in F_3 . They consist essentially of colloidal polysaccharides, the greater part of which was starch. The polysaccharides were determined by the Fehling solution after acid hydrolysis for 2.5 hours with a reflex condenser.

TABLE III
THE CARBOHYDRATES IN THE GREEN PLANT

	Roots		St	Stems		Branches		Leaves	
Material			June	3 collecti	on (1-4 i	nches)			
Total carbohydrates Reducing carbohydrates Non-reducing Polysaccharides	2.38	17.83 2.27 5.46 10.10	16.95 6.51 1.16 9.28	17.09 6.30 1.44 9.35			12.76 1.86 2.14 8.76	12.57 1.87 2.05 8.65	
	June 20 collection (6-8 inches)								
Total carbohydrates Reducing carbohydrates Non-reducing Polysaccharides	2.24 9.30	26.57 2.19 9.43 14.95	21.78 10.25 0.53 11.00	21.84 10.15 0.57 11.12			6.78 0.44 0.63 5.71	6.61 0.38 0.67 5.56	
			July	8 collect	ion (20 in	ches)			
Total carbohydrates Reducing carbohydrates Non-reducing	2.85	15.21 2.54 3.71 8.96	18.69 7.15 1.67 9.87	18.63 7.20 1.50 9.93	15.18 5.00 1.18 9.00	15.07 5.10 1.02 8.95	Trace 1.96 8.55	10.41 Trace 2.01 8.41	

The percentage of the different carbohydrates of various organs estimated at different times throughout the growth period is tabulated in table III. Curves showing the changes in different sugar content in these organs are given in figs. 6–8. These curves show that in the roots the reducing sugars remain constant, while the

non-reducing sugars fluctuate throughout the season. This is just the reverse of what is found in the stems. In the leaves the

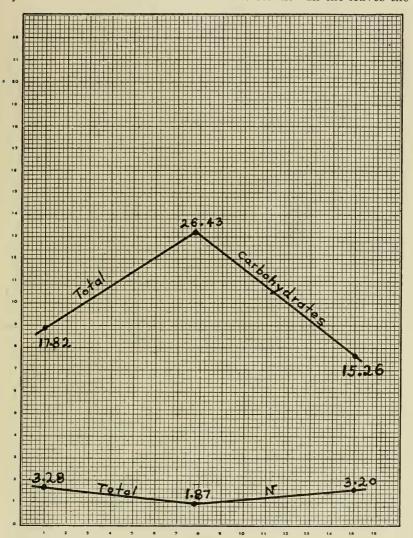


Fig. 9.—Reciprocal fluctuation of carbohydrate and nitrogen in roots (cf. figs. 3 and 6).

fluctuation of the reducing and non-reducing sugars is in the opposite direction; when the reducing sugars are high, the non-reducing

sugars are low, and the reducing type falls to zero at the time of seed formation.

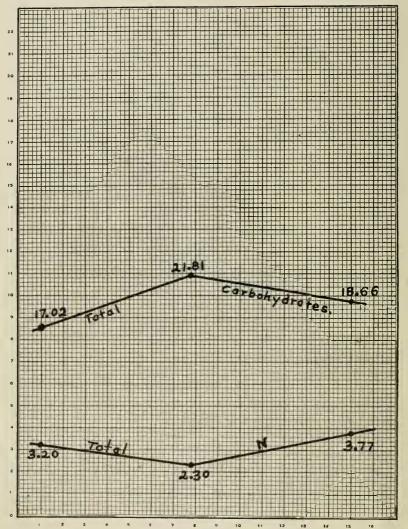


Fig. 10.—Reciprocal fluctuation of carbohydrate and nitrogen in stems (cf. figs. 4 and 7).

CARBOHYDRATE-NITROGEN RATIO.—According to the work of Kraus and Kraybill on the tomato, high nitrogen in a plant is

accompanied by low carbohydrate. "Whatever the conditions under which a plant has been grown, considering the whole plant

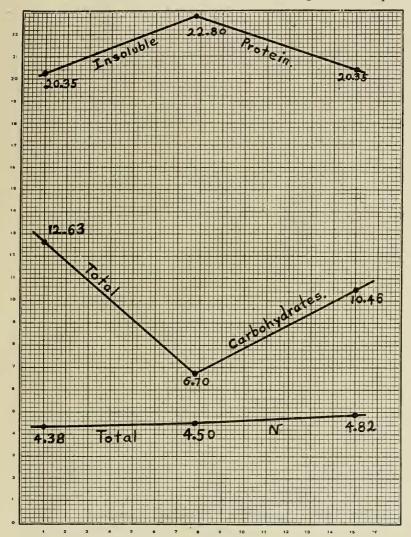


Fig. 11.—Reciprocal fluctuation of carbohydrate and nitrogen in leaves (cf. figs. 5 and 8); notice that carbohydrate reciprocates with insoluble protein and not with total nitrogen.

as a unit, increased total nitrogen and more particularly increased nitrate nitrogen are associated with increased moisture and decreased free-reducing substances, sucrose, polysaccharides, and total dry matter."

In the work with Amaranthus plants I have found a similar situation so far as the relation between nitrogen and carbohydrate is concerned; that is, low nitrogen is accompanied by high carbohydrate and high nitrogen by low carbohydrate. Upon computing the reciprocal condition in the different fractions I find that the product of carbohydrate by nitrogen is not a mathematical constant, but that it varies considerably, sometimes decreasing as the development of the plant progresses. The product varies least in the stem and roots.

Let C_1 , C_2 , and C_3 be the carbohydrates and N_1 , N_2 , and N_3 denote the nitrogen, the sub-numbers representing the time of collection. If the carbohydrate and nitrogen hold a reciprocal relation, then $\frac{C_1}{C_2} = \frac{N_2}{N_1}$, $\frac{C_2}{C_3} = \frac{N}{C_2}$, and $\frac{C_3}{C_1} = \frac{N_1}{C_3}$; by clearing the fractions, $C_1 \times N_1 = C_2 \times N_2 = C_3 \times N_3$, etc., or carbohydrate×nitrogen = constant K. Applying this principle, the following constants are obtained.

	Insoluble f	fraction (F ₃)		
	JUNE 3	June 20	JULY 8	Av. K.
Roots	15.7	13.06	10.92	13.23
Stems	16.20	10.62	9.20	12.01
Leaves	28.00	20.4	27.6	25.30
·	oluble fracti	ions (F ₁ +F ₂)		
Roots	10.10	11.00	6.27	9.12
Stems	,	8.97	7.88	8.43
Leaves	2.39	2.97	3.15	2.84

These data show that the carbohydrate-nitrogen ratio is not a constant as we think of a constant in mathematics or physics. In plants where great fluctuation occurs in their substratum throughout different parts of the day and different times in the season, this disparity is no positive evidence that such a ratio does not exist. Secondly, regardless of the exactness of the ratio, this much is true, when the carbohydrates are high the nitrogen compounds are relatively low, and vice versa. Figs. 9–11 show this reciprocal

condition in the roots, stems, and leaves in the various stages of development.

SEEDS

A tissue analysis of the seeds was made in order to discover what important compounds were present and the distribution of these compounds in the various fractions. This gives one a more comprehensive knowledge of the chemical constitution of the plant.

PREPARATION.—The seeds were freed from chaff and cleaned in a breeze until all red seeds were removed and only plump black ones remained. These uniform seeds were then ground in a mortar with the pestle by taking a few at a time. A known quantity (25 gm.) was weighed out in triplicate for alcoholic digestion, and the tissues were fractionated and the carbohydrates, the nitrogen compounds, and the phosphorus determined.

TABLE IV
SEEDS; SUMMARY OF TOTAL CONSTITUENTS

Material	A. bli	toides	A. retroflexus		
H ₂ O Av. 3 Total N Total P Total carbohydrates. Lipins Ash (ignition)	2.55 4.01 48.27 4.58	9.45 2.42 3.93 48.43 4.44 3.68	2.54 4.63 47.22 7.67 4.27	8.61 2.47 4.60 47.03 7.86 4.19	2·37 4·65 47·37 7·78 4·14

DIFFERENT FORMS OF PHOSPHORUS.—After separating the seeds into fractions F_1 , F_2 , and F_3 , the different phosphorus compounds were estimated in the three fractions. In each case the phosphorus was determined by the Pemberton-Kilgore method (10). The analysis of the different forms of phosphorus is given in table V, and the percentage ratio of these different forms to the total phosphorus is given in table VI. The inorganic phosphorus was estimated by the Chapin-Powick method (4). The amounts obtained from fractions two and three (F_2+F_3) were combined and calculated as total inorganic phosphorus. The soluble organic phosphorus was obtained by subtracting the inorganic phosphorus of fraction two only from total phosphorus in the same fraction (F_2) . The

phosphoprotein phosphorus was split off from the insoluble fraction (F₃) by digesting a weighed amount (1-3 gm.) with 1 per cent NaOH solution at 37-40° C. for 48 hours in an incubator, after which the solution was neutralized with acetic acid and made up to volume. The insoluble material was separated from the filtrate by the use of the centrifuge. The filtrate obtained was tested for phosphorus with magnesia mixture. The magnesium ammonium phosphate precipitated out by standing overnight was dissolved, and the phosphorus was reprecipitated as ammonium phosphomolybdate and then titrated against a standard alkali (8). The nucleoprotein phosphorus was estimated by the difference between the total phosphorus minus the phosphoprotein and the inorganic phosphorus in the two fractions (F₂+F₃). The lipin phosphorus was determined by taking an aliquot of the ether soluble fraction (F₁). The ether was first driven off on a steam bath before acid digestion and the percentage of this phosphorus was estimated in the same way.

Material	A. bli	toides	A. retroflexus		3
Inorganic P. Lipin P. Soluble organic P. Phosphoprotein P. Nucleoprotein P. Total P.	0.133	0.137	0.131	0.123	0.132
	0.014	0.013	0.019	0.020	0.017
	0.011	0.012	0.023	0.033	0.034
	1.240	1.340	1.840	1.950	1.790
	2.610	2.430	2.620	2.470	2.670
	4.008	3.932	4.633	4.596	4.649

TABLE VI
RATIO OF DIFFERENT P TO TOTAL P (PERCENTAGE P) IN SEEDS

Material	A. blitoides		les A. retroflexus		
Inorganic P. Lipin P. Soluble organic P. Phosphoprotein P. Nucleoprotein P. Total P.	0.35 0.28 31.00 65.30	3·49 0·33 0·31 34·10 61.80 100.00	2.83 0.41 0.49 39.74 56.60 100.07	2.65 0.44 0.72 42.40 53.70 99.91	2.84 0.37 0.73 38.50 57.50 99.94

Tables V and VI show the percentage of the different forms of phosphorus. The data in these tables show that about 96 per cent

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of the total phosphorus is in the organic combination in the seed, existing (perhaps) as phosphoprotein and nucleoprotein phosphorus. Both of these forms are insoluble in water, alcohol, ether, and alcohol-water solvents. The inorganic phosphorus is relatively low, and the writer believes that the figures given for inorganic phosphorus in table V are even too high, because the greater part of the inorganic phosphorus was obtained from the insoluble fraction F₃ (4). Moreover, there is no proof that the reagents used did not break down some of the organic phosphorus. The lipin phosphorus is very low, varying from 0.014 per cent in A. blitoides to 0.010 in A. retroflexus, calculated on dry weight basis. It is interesting to know that in all cases the different forms of phosphorus are relatively higher in A. retroflexus than the corresponding forms in A. blitoides.

TABLE VII DIFFERENT NITROGEN COMPOUNDS IN SEEDS (PERCENTAGE DRY WEIGHT)

Material	A. bl	itoides	A. retroflexus		S
Total N Nitrates NO ₃ Amino N Lipin N Soluble proteins Insoluble proteins Total proteins	2.550 0.200 0.096 0.027 2.260 12.640	2.420 0.212 0.095 0.027 2.270 12.450 14.720	2 . 540 0 . 194 0 . 089 0 . 031 2 . 660 12 . 700 15 . 360	2.470 0.193 0.090 0.032 2.790 12.820	2.370 0.205 0.090 0.033 2.890 12.250

TABLE VIII RATIO OF DIFFERENT N TO TOTAL N (PERCENTAGE DRY WEIGHT)

Material	A. blîtoides		A. retroflexus			
Total insoluble N		82.00 17.93 99.93	80.40 19.62 100.02	79.00 21.01 100.07	77.00 23.23 100.23	
Nitrate N. Lipin N. Amino N. Other soluble organic N	4.68	1.88 1.13 4.65 13.28	1.73 1.24 3.50 16.12	1.77 1.29 3.66 17.41	1.97 1.41 3.84 19.37	

NITROGEN COMPOUNDS.—The distribution of nitrogen in the different fractions of the seeds is about the same as that of phosphorus. The insoluble nitrogen comprised 80-83 per cent of the total nitrogen (tables VII and VIII). The soluble fractions contain only 17-20 per cent of the total, and most of it exists as organic nitrogen. The portion representing inorganic nitrogen is the nitrate nitrogen, which is relatively small. Calculated as nitrates (NO₃), the seeds contain 0.20 per cent, and this is equivalent to 1.80 per cent of the total nitrogen (tables VII and VIII). The lipin nitrogen is very small, only 0.027 per cent in A. blitoides and 0.032 per cent in A. retroflexus. These represent 1.10 and 1.31 per cent respectively of the total nitrogen content in these two seeds. In general a high percentage of insoluble phosphorus is accompanied by a high percentage of insoluble nitrogen, and a low percentage of soluble phosphorus by a low percentage of soluble nitrogen.

CARBOHYDRATES.—The polysaccharides are the predominating sugars in these seeds. A. retroflexus seeds contain 46 per cent and A. blitoides 47.75 per cent polysaccharides (on dry weight

TABLE IX

CARBOHYDRATES IN SEEDS (PERCENTAGE DRY WEIGHT)

Material	A. bl	itoides	A. retroflexus			
Lipin sugars	None	None	None	None	None	
	None	None	None	None	None	
	0.67	0.68	1.12	1.13	1.17	
	47.60	47.75	46.10	45.90	46.20	
	48.27	48.43	47.22	47.03	47.37	
	Ratio of different sugars to the total carbohydrates					
Non-reducing	1.38	1.40	2·37	2.40	2.47	
	98.65	98.60	97.60	97.60	97.60	
	100.03	100.00	99·97	100.00	100.07	

basis). If these sugars are calculated on the dry basis of the total sugars, the polysaccharides represent 97.60 and 98.60 per cent respectively in these two species. A striking contrast is seen on comparing the amount of polysaccharides in the green plant organs (figs. 6–8), which vary only slightly throughout the growing period, with that found in the seeds. In the growing period the highest percentage of polysaccharides was only 14.85, while that of the seeds was 47. In addition to this noticeable contrast, the soluble

sugars, both reducing and non-reducing, were comparatively high in the green plant organs (6-8 per cent), while those of the seeds are low (0.67-1.14 per cent).

LIPIN FRACTION.—The percentage of this fraction is 4.5 for A. blitoides and 7.78 for A. retroflexus. A closer examination of this fatlike substance shows that it contains phosphorus and nitrogen. The percentage of this nitrogen and phosphorus is very low (calculated on dry weight basis of whole sample). The presence of nitrogen and phosphorus indicates that the seeds contain phosphotides. The atomic ratio of nitrogen to phosphorus was determined by dividing the percentage by their respective atomic weights. The atomic ratio of N:P for A. blitoides is 1:2.3, and that for A. retroflexus is 1:2.6. This shows that other forms of phosphorus must be present than that existing in "ideal lecithin," which implies that the atomic ratio of N:P=1:1 (5, 13).

In addition to lecithin, a phytosterol was present in the seeds. This could not be quantitatively determined by using animal cholesterol as a standard, because animal cholesterol (17), has a different tint from that of the plant. Buchard's color reaction gives a deep blue color for animal cholesterol, while for the seeds it gives a yellowish green. The amount of phytosterol in these two species of seeds was compared. Assuming that the phytosterol in A. blitoides is unity, that in A. retroflexus is 2.8.

TABLE X CONSTITUENTS OF LIPIN FRACTION (F1) OF SEEDS (PERCENTAGE DRY WEIGHT)

Material	A. blitoides			A. retroflexus			
$ \begin{array}{c} \text{Lipin, fats, etc. (wet basis)}.\\ \text{Lipin (dry basis)}.\\ \text{P in } F_1.\\ \text{P calculated in whole sample}\\ \text{N in } F_1.\\ \text{N calculated in whole sample} \end{array} $	4.15 4.58 0.300 0.014 0.600 0.027	4.07 4.44 0.300 0.014 0.640 0.027	4.23 4.66	7.02 7.67 0.240 0.019 0.402 0.031	7.18 7.86 0.260 0.020 0.400 0.031	7.09 7.78 0.220 0.017 0.420 0.033	

INORGANIC ELEMENTS.—The 1917 seeds of Amaranthus retroflexus were used for the estimation of the inorganic elements. The percentage of total ash given in table IV is 3.50 per cent for A. blitoides and 4.20 for A. retroflexus, but this is actually too low for the inorganic elements present in the seeds, because the total phosphorus alone is 4.0 and 4.60 per cent respectively for the two species. This discrepancy is due perhaps to the loss of nitrates and part of the sodium, potassium, etc., in burning in the electric muffle. In the following analysis of the inorganic elements acid digestion (concentrated H_2SO_4 +concentrated HNO_3) was used for the kations and alkaline fusion for the anions. The potassium was determined directly in the presence of all other elements except ammonia (NH_4 -ion) and strong acid, as K_2NaCO (NO_2)₆- $\frac{1}{2}$ H_2O (11). Magnesium was estimated by the volumetric method as NH_4MgAsO_4 (7). The percentage of inorganic elements is as follows:

W 5 .						
	INORGANIC	SALTS	(A retr	oflexus);	1917 SEEDS	
	,				Percentage	Percentage
Silica	<i></i>				0.42	0.40 °
$Al_2O_3+F_2O_3$	Эз				0.53	0.56
CaO					0.54	. 0.53
MgO					0.82	o.84
K₂O					0.38	0.35
Ńa₂O					no o	letermination
Cl					Trace	Trace
SO_3					0.34	0.33
P_2O_5					8.90	8.85
N ₂ O ₃ (nitr	ates)				O.I2	0.123
Total	(not include	ding N	a ₂ O)		12.05	11.98

Discussion

In spite of the inaccuracy of Boutin's method for the determination of the nitrates, his results are probably not far from correct. He stated his results in percentage of KNO₃ as shown in the second column of table XI. I have calculated these as percentage of NO₃, as shown in the third column. His percentages of nitrates are not far from those found by me for the stems (8.57 per cent) and branches (12.50 per cent) of A. retroflexus, July 8 collection, as shown in table II.

	Percentage KNO	Percentage equivalent in NO
A. retroflexus	22.77	13.92
A. blitum	11.68	7.17
A. ruber	16.0	9.82

Table XI shows the results of the analyses of the seeds by various authors. In the main there is fairly close agreement, but in some cases there are considerable discrepancies. The discrepancies can probably be explained by the different chemical methods used by the various authors and by the lack of uniformity in the different crops analyzed.

 $\begin{tabular}{ll} TABLE~XI \\ Comparison~of~some~of~the~analyses~on~Amaranthus~seeds \\ \end{tabular}$

	PAMMEL	PAMMEL AND DOX		HARDING AND EGGE			Woo	
MATERIAL		A. retro- flexus	A. retroflexus				A. retro-	
	A. blitoides		20 mesh	72 mesh	Oven dry	A. blitoides	flexus	
Condition of seeds			Non-	uniform		Matured uniform	Matured uniform	
H ₂ O				11.28	8.60	Av. of 3	Av. of 3 8.61	
Lipins (fats)	Little*	Little*		7.92	8.46	4.56 47.68	7.77	
Reducing sugars				Trace	Trace	None None	47.21 None	
Non-reducing sugars (sug			2.08	2.15	2.35	0.67	1.14	
Nitrogen	г.88	2.49				2,48	2.46	
Protein		15.59	18.57	19 13	4.88	3.59	15.03	

^{*} Microchemical test.

From the results of this study it would seem that Amaranthus retroflexus, and probably other species of the same genus, can bear, as they ordinarily do bear, large amounts of free nitrates without being forced out of reproduction into extreme vegetation. This genus apparently is endowed with a very high capacity for nitrate absorption, as well as for maintaining its full seed production power in the face of a great excess of free nitrates. In this respect it seems to differ from the tomato studied by Kraus and Kraybill, and probably from many other plants. Considering all angiosperms, it is likely that, due to hereditary characters, there is a great range of ease with which plants can be forced to excessive vegetation by extreme nitrate supply within the plant. It is well known that a given level of fertility that will throw small grains into extreme straw production with deficiency of grain will give excellent grain production in corn. This may be due to the lower nitrate absorbing power of the corn, to its greater

photosynthetic activity to balance the nitrates absorbed, or to the higher carbohydrate-nitrogen ratio accompanying best grain production. Which of these three possibilities really determines the situation can only be answered by such studies as those made or suggested on the tomato by Kraus and Kraybill, or studies of the type made in this paper. It is evident, however, that there is need of numerous studies of the carbohydrate-nitrogen ratio in plants, both in regard to the factors affecting this ratio and the effect of the ratio on plant characters. As was suggested in the review of the literature at the beginning of this article, such studies are likely to throw much light on other physiological features than vegetation and reproduction.

Summary

- 1. There is a large amount of nitrate in the organs of A. retro-flexus. The stem and branches are the primary nitrate storage organs. The rate of nitrate absorption increases with the aging of the plant, perhaps partly being due to the development of the root system with numerous branching rootlets, increasing the radius of the feeding area from a few inches to 2 ft. or more.
- 2. This high capacity for nitrate absorption and storage must be an important factor in making *Amaranthus* a very successful competitor against cultivated plants, so effectively withdrawing as it does the nutrient element most commonly limiting plant production. It would be interesting to know how generally and to what degree weeds possess this power.
- 3. The carbohydrates and nitrogen compounds fluctuate throughout the growing period. The fluctuation of the carbohydrates is in the reverse order of the nitrogen compounds. This inverse ratio is not a truly mathematical constant, but in general when the carbohydrates are high the nitrogen compounds are low, and vice versa. As the nitrate nitrogen composes more than 50 per cent in the stems and branches, there is a possibility that nitrates have some modifying effects on this reciprocal relationship. This inverse ratio is due partly to the synthesis of protein, chlorophyll, phospholipin, and other organic nitrogen compounds at the expense of the soluble carbohydrates.

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- 4. Tissue analysis of the seeds shows the distribution of different forms of phosphorus in the various fractions. The organic phosphorus, which consists chiefly of phosphoprotein and nucleoprotein phosphorus, is high, and that of the inorganic form is low.
- 5. The distribution of nitrogen in seeds is in the same order as that of the phosphorus. The insoluble portion contains 80-83 per cent of the total. The soluble part varies from 17 to 20 per cent, most of which is in the organic form. The inorganic form is represented by the nitrate nitrogen.
- 6. The predominating sugars in the seeds are the polysaccharides. These compose nearly one-half of the total dry weight of the seeds. In both A. retroflexus and A. blitoides there is absence of lipin sugars in F₁ and reducing sugars in F₂. Only a small amount of non-reducing sugars was present in the two varieties.
- 7. The presence of nitrogen and phosphorus in the lipin fraction indicates that the seeds contain phosphatides. Phytosterol was also present. By comparison, A. retroflexus has 2.8 times as much as A. blitoides.

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UNIVERSITY OF CHICAGO

LITERATURE CITED

- 1. BOUTIN, A., Sur la présence d'une proportion considérable de nitre dans l'Amaranthus blitum. Compt. Rend. 76:413-417. 1873.
- 2. ———, Sur le présence d'une proportion considérable de nitre dans deux variétés d'Amaranthus. Compt. Rend. 78:261-262. 1874.
- 3. Brosset, —, Sur quelques passages de Stan. Bell, d'où l'on peut conclure que l'Amaranthus blitum est cultivé en circassie pour le nitre qu'il contient. Compt. Rend. 70:1274. 1874.
- 4. CHAPIN, R. M., and POWICH, W. C., An improved method of the estimation of inorganic phosphoric acid in certain tissues and food products. Jour. Biol. Chem. 20:97-114. 1915.
- 5. CZAPEK, F., Biochemie der Pflanzen. 2d ed. 1:763-773. 1913.
- 6. FISCHER, H., Zur Frage der Kohlensame-Ernährung der Pflanzen. Gartenflora 65:232-237. 1916.
- 7. Fox, Paul I., Titration of calcium and magnesium in the same solution. Jour. Ind. Eng. Chem. 5:010-013. 1013.

- 8. HARDING, E. P., and EGGE, W. A., A proximate analysis of seed of common pigweed, *Amaranthus retroflexus* L. Jour. Ind. Eng. Chem. 10:529-530. 1918.
- 9. Hedlund, T., Über die Möglichkeit, von der Ausbildung des Weizens in Herbst auf die Winterfestigheit der verschiedenen sorten zu schliessen. Rev. Bot. Centralbl. 135:222-224. 1917.
- 10. HIBBARD, P. L., A study of the Pemberton-Kilgare method for determination of phosphoric acid. Jour. Ind. Eng. Chem. 5:998-1009. 1913.
- II. KOCH, W., Methods for quantitative chemical analysis of animal tissue. Jour. Amer. Chem. Soc. 31:1329-1364. 1909.
- 12. Kraus, E. J., and Kraybill, H. R., Vegetation and reproduction with special reference to the tomato. Oregon Agric. Exp. Sta. Bull. 149. pp. 90. 1918.
- 13. MacClean, Hugh, Lecithin and allied substances, the lipines. Monograph Biochemistry, pp. vii-206. 1908.
- Official and provisional methods of analysis. Association of Official Agricultural Chemists. Bull. 107. Bur. Chem., U.S. 1907.
- 15. Pammel, H. L., and Dox, A. W., The protein contents and microchemical tests of the seeds of some common Iowa weeds. Proc. Iowa Acad. Sci. 22:527-532. 1917.
- 16. RIBERA, U., Über die Ursache des Lagems beim Weizen. Internat. Agric. Techn. Rundschau 7:524-525. 1916.
- 17. Weston, P. G., Colorimetric methods for determining serum cholesterol. Jour. Biol. Chem. 28:383-387. 1917.